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152 / Chapter 21 / 1 /  
Radiation and  
Radioactive Isotopes  
Chapter 21—Nuclear  
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~~Chemistry: Part 3 of 9~~

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~~Carbon-13 NMR~~

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#39 Einstein's Proof of

$E=mc^2$  How to

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Defect and Binding

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1 of 17 #Q value of a

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es Half Life Chemistry  
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Radioactive Decay  
Calculations Practice  
Examples

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Nuclear Chemistry,  
'Alpha Decay' by Dr.  
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III, Chemistry Chapter  
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Chemistry SECTION 4

SHORT ANSWER

Answer the following questions in the space provided. 1. Match each

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of the following  
statements with the  
process(es) to which  
they apply, using one of  
the choices below: (1)  
fission only (3) both  
fission and fusion

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Answers The n:p ratio for Cr-53 is  $\frac{29}{24} = 1.21$ ; for Mn-51, it is  $\frac{26}{25} = 1.04$ ; for Fe-59, it is  $\frac{33}{26} = 1.27$ . Positron decay occurs when the

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nucleons. protons and

neutrons. nuclide. An

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atom identified by the  
number of protons

## Answers

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Transformations •

Rutherford in 1919

performed the first  
nuclear transformation.

- The transmutations are sometimes represented

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by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation

becomes:  ${}_{14}^{27}\text{N} + {}_2^4\text{He} \rightarrow {}_{17}^{31}\text{O} + {}_1^1\text{H}$

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fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction. The radioactive isotope is contained in tubes called fuel rods. A moderator. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear

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reactions. A coolant.

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